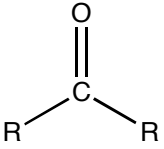
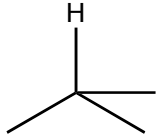
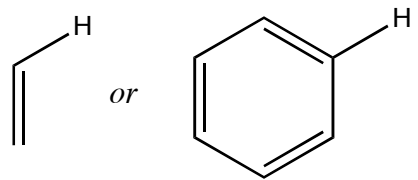
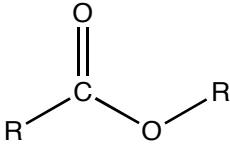
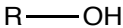
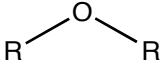
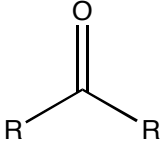
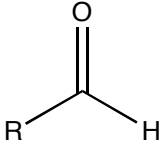
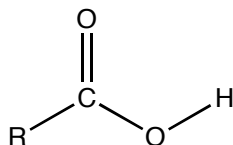


<p>Approximate IR absorption area for a carbonyl group:</p> 	<p>Absorption in the IR spectrum at $\sim 1700 \text{ cm}^{-1}$</p>
<p>Approximate IR absorption area for a hydroxyl group:</p> <p>R—OH</p>	<p>Absorption in the IR spectrum at $\sim 3500 \text{ cm}^{-1}$</p>
<p>Approximate IR absorption area for saturated CHs:</p> 	<p>Absorption in the IR spectrum at $< 3000 \text{ cm}^{-1}$</p>
<p>Approximate IR absorption area for unsaturated CHs:</p> 	<p>Absorption in the IR spectrum at $> 3000 \text{ cm}^{-1}$</p>
<p>Approximate IR absorption area for a CO single bond:</p> <p>C—O</p>	<p>Absorption in the IR spectrum at $\sim 1000 - 1250 \text{ cm}^{-1}$</p>

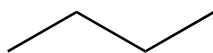
<p>Expected IR absorptions for an ester molecule:</p> 	<p>Class of molecule has IR absorptions at: $\sim 1700 \text{ cm}^{-1}$ $\sim 1000 - 1250 \text{ cm}^{-1}$ (plus saturated and/or unsaturated CH)</p>
<p>Expected IR absorptions for an alcohol molecule:</p> 	<p>Class of molecule has IR absorptions at: $\sim 3500 \text{ cm}^{-1}$ $\sim 1000 - 1250 \text{ cm}^{-1}$ (plus saturated and/or unsaturated CH)</p>
<p>Expected IR absorptions for an ether molecule:</p> 	<p>Class of molecule has IR absorptions at: $\sim 1000 - 1250 \text{ cm}^{-1}$ (plus saturated and/or unsaturated CH)</p>
<p>Expected IR absorptions for a ketone molecule:</p> 	<p>Class of molecule has IR absorptions at: $\sim 1700 \text{ cm}^{-1}$ (plus saturated and/or unsaturated CH)</p>
<p>Expected IR absorptions for an aldehyde molecule:</p> 	<p>Class of molecule has IR absorptions at: $\sim 1700 \text{ cm}^{-1}$ (plus saturated and/or unsaturated CH)</p>

Expected IR absorptions for a carboxylic acid molecule:



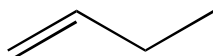
Class of molecule has IR absorptions at:
Very broad absorption at $\sim 3000\text{cm}^{-1}$
 $\sim 1700\text{cm}^{-1}$
 $\sim 1000 - 1250\text{cm}^{-1}$
(plus saturated and/or unsaturated CH)

Expected IR absorptions for an alkane molecule:



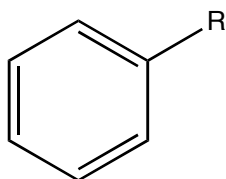
Class of molecule has IR absorptions at:
 $< 3000\text{cm}^{-1}$

Expected IR absorptions for an alkene molecule:



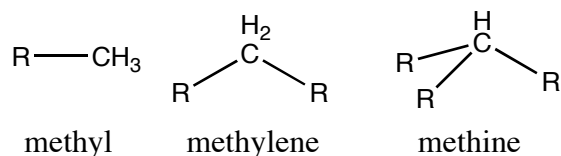
Class of molecule has IR absorptions at:
 $> 3000\text{cm}^{-1}$
(often saturated CH too)

Expected IR absorptions for an aromatic molecule:



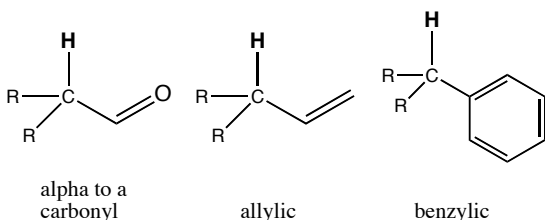
Class of molecule has IR absorptions at:
 $> 3000\text{cm}^{-1}$
(often saturated CH too)

Approximate ^1H NMR chemical shift area for a simple saturated C–H:



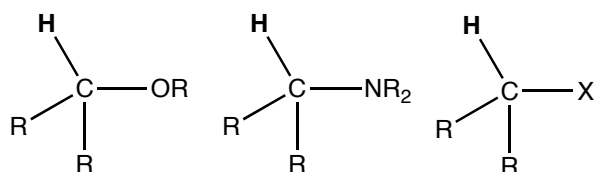
The type of C–H that appears in the 0 – 1.5 PPM range in the ^1H NMR spectrum

Approximate ^1H NMR chemical shift area for:



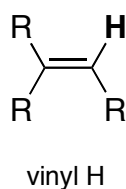
The types of C–H that appear in the 1.5 – 2.5 PPM range in the ^1H NMR spectrum

Approximate ^1H NMR chemical shift area for:



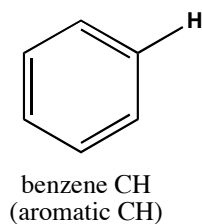
The types of C–H that appear in the 2.5 – 4.5 PPM range in the ^1H NMR spectrum

Approximate ^1H NMR chemical shift area for:



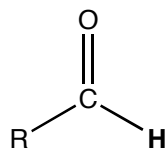
The type of C–H that appears in the 4.5 – 6.5 PPM range in the ^1H NMR spectrum

Approximate ^1H NMR chemical shift area for:



The type of C–H that appears in the 6.5 – 8.5 PPM range in the ^1H NMR spectrum

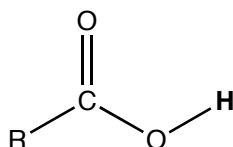
Approximate ^1H NMR chemical shift area for:



aldehyde

The type of C–H that appears in the 9 – 10 PPM range in the ^1H NMR spectrum

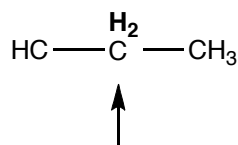
Approximate ^1H NMR chemical shift area for:



carboxylic acid

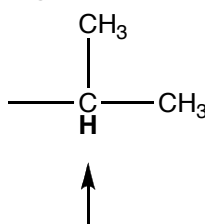
The type of C–H that appears in the 10 – 12 PPM range in the ^1H NMR spectrum

What is the multiplicity of the following set of protons?



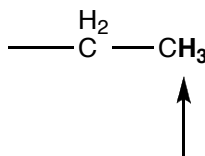
How many neighbors would a methylene group have that is a **pentet** in the ^1H NMR spectrum?

What is the multiplicity of the following set of protons?



How many neighbors would a methine group have that is a **heptet** in the ^1H NMR spectrum?

What is the multiplicity of the following set of protons?



How many neighbors would a methyl group have that is a **triplet** in the ^1H NMR spectrum?