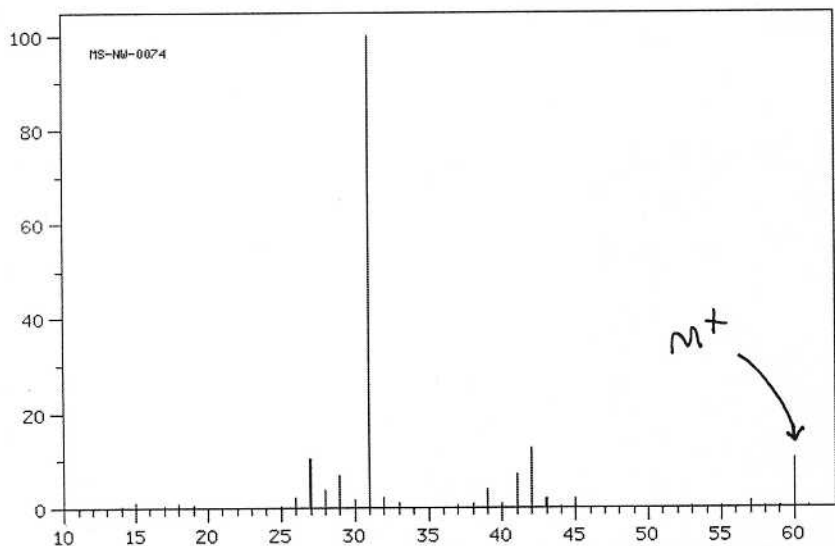
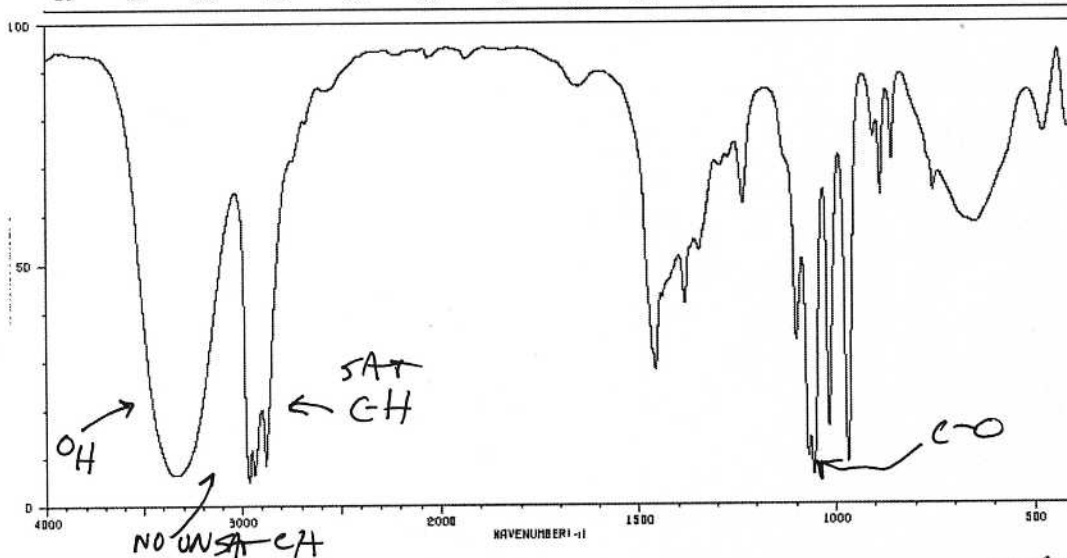


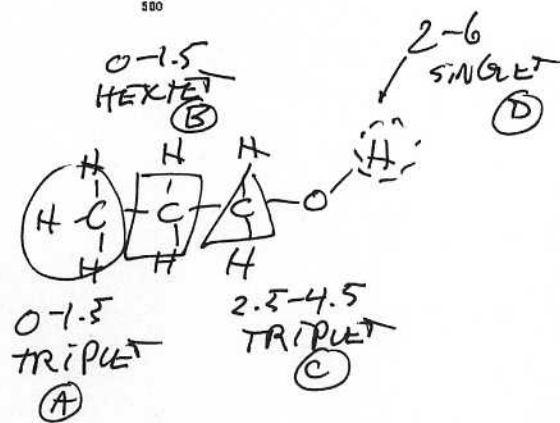
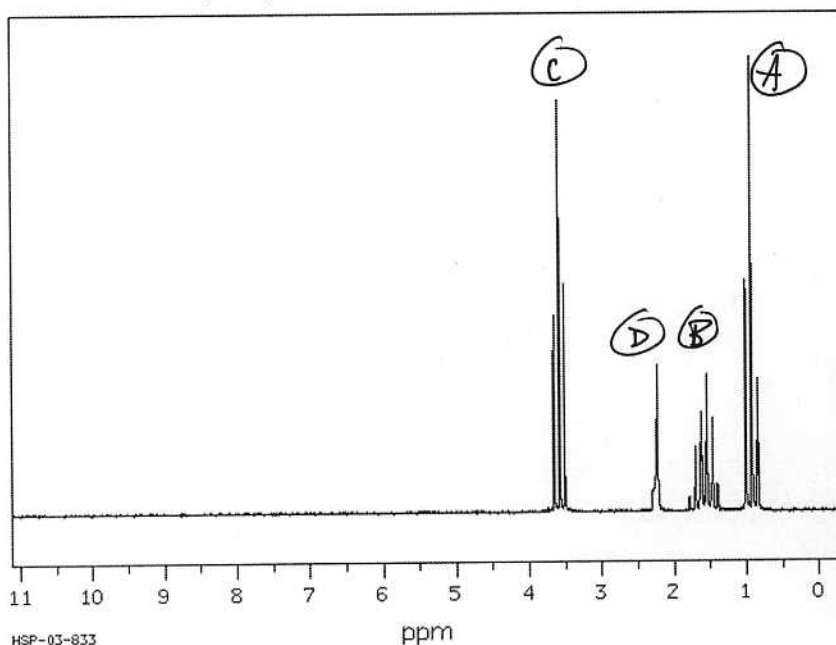
Spectroscopy Problem 13



$MASS = 60$
 NO HALOGENS
 $\frac{60}{14} = 4$ HEAVY ATOMS



\therefore ALCOHOL
 $R-OH$
 $R = 3$ CARBONS
 $+ O-H$
 $= 4$ HEAVY ATOMS



C_3H_8O
 $MW = 60$ ✓