## **Chapter 23 Practice Problems**

1. Identify each of the following amines as  $1^{\circ}$ ,  $2^{\circ}$  or  $3^{\circ}$ 

2. The basicity of amines requires the nitrogen atom to donate a lone pair of electrons to a proton source as shown below:

- A. Given this information, explain why p-chloroaniline is a weaker base than aniline.
- B. Rank the following compounds according to their expected basicities:

## 3. Practice the following problems for the synthesis of amines by providing the missing items as required:

$$F.$$
 $O$ 
 $N$ 
 $N$